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# Influence of solvent evaporation on ultimate tensile strength of contemporary dental adhesives

Atsushi Kameyama<sup>1,2\*</sup> , Akiko Haruyama<sup>2</sup> , Hirokazu Abo<sup>3</sup>, Masashi Kojima<sup>2</sup>, Yuichi Nakazawa<sup>2</sup> and Takashi Muramatsu<sup>2</sup>

\*Correspondence:  
atsushi.kameyama@mdu.  
ac.jp

<sup>1</sup> Department of Operative Dentistry, Endodontology, and Periodontology, School of Dentistry, Matsumoto Dental University, 1780 Gobara Hirooka, Shiojiri, Nagano 399-0781, Japan  
Full list of author information is available at the end of the article

## Abstract

The aim of this study was to evaluate the influence of solvent evaporation on the ultimate tensile strength (UTS) of commercial adhesives. Two 1-step (OptiBond All-In-One and G-Premio Bond) and two 2-step (Clearfil SE Protect, OptiBond XTR) adhesives were selected. Two bottles of each adhesive were opened and stored at 37 °C in a dry oven with silica gel shielded from light for 2 weeks (“Desiccated”). Two unopened bottles were stored at room temperature (“Original”). After 2 weeks, the adhesives were used to fill an hour-glass shaped, metallic matrix mold and light-cured. Samples were weighed, and then immersed in a 37 °C water bath for 1 h or 7 days. The UTS of each sample was then measured at a cross-head speed of 1 mm/min ( $n = 10$ ). The UTS for the Clearfil SE Protect was higher in the “Original” than “Desiccated” samples ( $p < 0.05$ ). For the OptiBond XTR, no significant difference was found between the ‘Original’ and ‘Desiccated’ samples ( $p > 0.05$ ). Neither of the two “Original” 1-step samples could be hardened, even after light-curing, yet the ‘Desiccated’ OptiBond All-In-One samples obtained high UTS values. Both OptiBond All-In-One and Clearfil SE Protect had an increase in weight after the 7-day immersion in water. In conclusion, residual solvent reduces the mechanical strength of the adhesive. The hydrophilicity of the adhesive resin might also affect its mechanical strength.

**Keywords:** Dental adhesive, Solvent evaporation, Ultimate tensile strength, Water sorption, HEMA

## Introduction

Dental practice has been significantly revolutionized by the development of composite filling materials [1]. These materials not only satisfy the esthetic demands of the patients, but also allow for easy manipulation and contouring of the material to replace or fill the lost or damaged tooth. Additionally, the use of a dental adhesive is much less invasive and conserves the tooth structure, with minimal visibility. A resin-based composite filling material can be used routinely in the clinic to restore caries and worn structures, close diastemas, and remodel malformed or fractured teeth [2]. It is essential that a durable bond is formed between the adhesive and the dental tissue during direct composite restoration. Therefore, adhesive resin monomers should properly integrate and polymerize with the demineralized dental tissue.

Contemporary adhesives have been categorized as “etch-and-rinse” and “self-etch” [3]. Among them, the two-step self-etch adhesives (2-SEAs) are popular because of their ease of use and rapid application [4]. Self-etching primers generally contain acidic functional monomers that are dissolved in water. Therefore, they do not require a separate rinse step after etching, as the acidic and hydrophilic functional monomer simultaneously demineralizes and infiltrates the enamel/dentin substrate [5]. These hydrophilic monomers promote wettability and infiltration of the adhesive resin, which is mainly composed of hydrophobic resin monomers; for example, 2,2-bis[4-(2-hydroxy-3-methacryloyloxypropoxy)phenyl]propane (Bis-GMA), urethane dimethacrylate (UDMA), and triethylene glycol dimethacrylate (TEGDMA).

Recent formulations have shifted to one-step self-etch adhesive (1-SEA) systems, in which all components are combined into a single solution. Despite their user-friendliness, the lack of a hydrophobic bonding resin in these 1-SEA formulations reduces the stability of the bond over time, because the bonded interfaces behave as semipermeable membranes that allow the movement of water across them, expediting hydrolytic degradation [6, 7].

Since almost all 1-SEA reagents contains solvents such as water, ethanol or acetone, evaporation of the solvent affects monomer conversion and the mechanical properties of the agent after setting [8, 9]. However, some 2-SEA agents have a solvent as part of the ingredients of the bonding agent, and some 1-SEA and 2-SEA products contain 2-hydroxyethyl methacrylate (HEMA), a hydrophilic compound that can affect water absorption by the cured adhesive [10].

Therefore, the purpose of this study was to evaluate the influence of solvent evaporation on the ultimate tensile strength (UTS) of commercial adhesives. The null hypotheses of this study were that (1) residual solvent does not influence the mechanical properties of the cured adhesive regardless of the period of water storage, and (2) mechanical properties of the cured adhesive are influenced by the presence of HEMA regardless of the period of water storage.

## Methods

### Study design and adhesives tested

This study followed a factorial 4×2×2 design for “adhesive”, desiccation” and “storage period”.

Two 2-SEAs (Clearfil SE Protect, Kuraray Noritake Dental, Tokyo, Japan; OptiBond XTR, Kerr, Orange, CA, USA) and two 1-SEAs (OptiBond All-In-One, Kerr; G-Pre-mio Bond, GC, Tokyo, Japan) were used in this study (Table 1). For each adhesive, four bottles with the same batch number were used and randomly divided into two groups (“Original” and “Desiccated”).

### Active solvent evaporation for “Desiccated” group

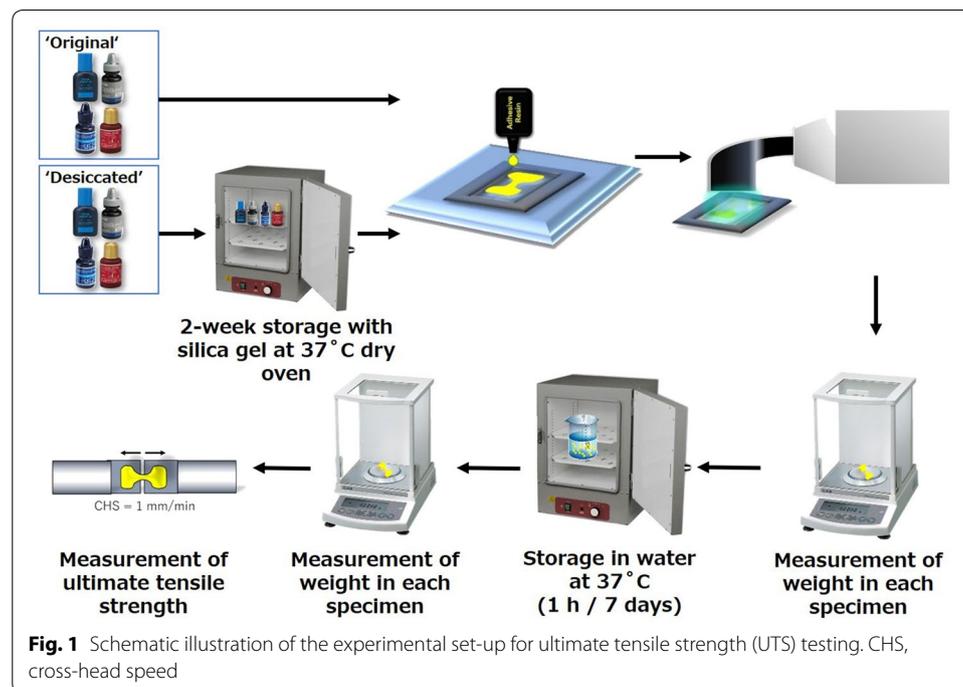
Two bottles of each adhesive were opened and the nozzle removed using pliers in a dark room. The adhesives were stored in a dark dry oven at 37 °C with silica gel (Wako Pure Chemical, Osaka, Japan) shielded from light for 2 weeks to aggressively evaporate any residual solvent. After 2-weeks, the nozzle and the bottle cap were replaced and the adhesive was stored at a controlled temperature (25 °C) until specimen preparation. The

**Table 1** Materials used in this study

Adhesive (manufacturer)	Ingredients	Batch no.
2-step self-etch adhesive		
Clearfil SE Protect (Kuraray Noritake Dental) <sup>a</sup>	Bis-GMA, HEMA, 10-MDP, Hydrophobic aliphatic methacrylate, colloidal silica, CQ, initiators and accelerators, sodium fluoride, others	CK0014
OptiBond XTR (Kerr)	Hydrophobic monomers, ethanol, camphorquinone, 0.4- $\mu$ m barium glass, nano-silica filler, sodium hexafluorosilicate	5143490
1-step self-etch adhesive		
OptiBond All-In-One (Kerr)	GPDM, glycerol dimethacrylate, HEMA, water, acetone, ethanol, CQ-based photo-initiator, nano-sized fillers, sodium hexafluorosilicate, ytterbium fluoride	5164557
G-Premio Bond (GC)	4-MET, MDP, MDTP, methacrylate monomer, acetone, water, photo initiators, silica	1605241

Bis-GMA, 2,2-bis[4-(2-hydroxy-3-methacryloyloxypropoxy)phenyl]propane; CQ, *dl*-camphorquinone; GPDM, glycerol-phosphate-dimethacrylate; HEMA, hydroxyethylmethacrylate; 4-MET, 4-methacryloyloxyethyl trimellitate; 10-MDP, methacryloyloxydecyl dihydrogen phosphate; MDTP, methacryloyloxydecyl dihydrogen thiophosphate

<sup>a</sup> Also known as "Clearfil Megabond FA" in Japan



other two bottles were stored at a controlled temperature (25 °C) until specimen preparation with the bottle cap sealed.

#### Measurements of specimen weight and ultimate tensile strength (UTS)

Figure 1 shows a schematic representation of specimen preparation and the method of UTS testing. A stainless steel, hour glass-shaped split mold of 1 mm thickness (Fig. 2) was constructed and fixed onto a glass slide. The mold was then filled with adhesive, pressed with a second 1-mm-thick glass slide, and then light cured for 60 s through the top glass slide using an LED light-curing unit (Demi Plus, Kerr) [11, 12]. Light intensity was controlled using a hand-held dental radiometer (Model L.E.D.

Radiometer, SDS Kerr, Middleton, WI, USA) to ensure a light output of at least 1000 mW/cm<sup>2</sup> [13]. After light curing, the samples were carefully removed from the mold. The flash was removed with a surgical blade #15C. The weight of each specimen was immediately measured using a digital analytical balance (HR-202i, A&D Co., Tokyo, Japan). The specimens were then randomly divided and assigned to one of two storage conditions: 1 h or 7 days in 37 °C water bath shielded from light. After incubation, each specimen was weighed again, and the cross-sectional area at the narrowest part of the specimen was measured using a digital caliper (Mitutoyo, Tokyo, Japan) to the nearest 0.01 mm. The remaining specimens were attached to the two free-sliding parts of a purpose-built holding device (Micro Tensile Test Jaw, Bisco, Schaumburg, IL, USA) using cyanoacrylate glue (Model Repair II Blue, Dentsply-Sankin, Ohtawara, Tochigi, Japan), and a tensile load was applied with a Micro Tensile Tester (Bisco, Schaumburg, IL, USA) at a cross-head speed of 1 mm/min until the specimen fractured ( $n = 10$ ). The UTS of the cured adhesive was calculated as:

$$\text{UTS} = F/A \quad (1)$$

where  $F$  is the tensile load force at failure (N), and  $A$  is the cross-sectional area of the narrowest part of the specimen (mm<sup>2</sup>). The ultimate tensile strength (N/mm<sup>2</sup>) was expressed in MPa [11, 12].

### Statistical analysis

Data were analyzed using a three-way analysis of variance (ANOVA). If differences were found, pair-wise testing was performed using Tukey's HSD test. The significance level was set to  $\alpha = 0.05$ . Statistical analysis was performed using IBM SPSS 18 statistical software (SPSS, Chicago, IL, USA).

## Results

### Ultimate tensile strength

The result for the UTS tests is shown in Table 2. No effect was found for "storage period" ( $p = 0.072$ ), whereas a significant effect was found for "adhesive" ( $p = 0.000$ ) and "desiccation" ( $p = 0.000$ ). A significant interaction was also found for the three factors ( $p = 0.000$ ).

Neither of the 1-SEAs (OptiBond All-In-One and G-Premio Bond) hardened in their "Original" states, even after sufficient light exposure. Therefore, these results were eliminated for statistical analysis.

The UTS of the "Desiccated" OptiBond All-In-One was  $35.6 \pm 15.2$  MPa after 1 h and  $28.2 \pm 6.2$  MPa after 7 days, respectively. In contrast, G-Premio Bond showed the significantly lower UTS values than OptiBond All-In-One at both time points ( $p < 0.05$ ).

Differ to 1-SEAs, both 2-SEAs hardened in their "Original" and "Desiccated" forms. The UTS of the "Original" Clearfil SE Protect was significantly higher than the "Desiccated" adhesive samples at 1 h ( $43.7 \pm 11.6$  vs.  $29.9 \pm 9.9$ , respectively;  $p = 0.008$ ); in contrast, by 7 days, "Desiccated" group was significantly higher than the "Original" group ( $24.8 \pm 6.8$  vs.  $36.6 \pm 6.9$ , respectively;  $p = 0.002$ ). No significant differences were found between the "Original" and "Desiccated" OptiBond XTR samples ( $p = 0.298$ ).

**Table 2 Ultimate tensile strengths**

Group	Adhesive	1 h	7 days
Original	Clearfil SE Protect	43.7 ± 11.6 <sup>A,a</sup>	24.8 ± 6.8 <sup>B,b</sup>
	OptiBond XTR	16.5 ± 4.3 <sup>B,b</sup>	20.2 ± 7.6 <sup>B,a</sup>
	OptiBond All-In-One	NA*	NA*
	G-Premio Bond	NA*	NA*
Desiccated	Clearfil SE Protect	29.9 ± 9.9 <sup>B,b</sup>	36.6 ± 6.9 <sup>A,a</sup>
	OptiBond XTR	25.0 ± 7.5 <sup>BC,a</sup>	23.8 ± 9.7 <sup>B,a</sup>
	OptiBond All-In-One	35.6 ± 15.2 <sup>AB</sup>	28.2 ± 6.2 <sup>AB</sup>
	G-Premio Bond	6.8 ± 1.9 <sup>D</sup>	7.1 ± 1.4 <sup>C</sup>

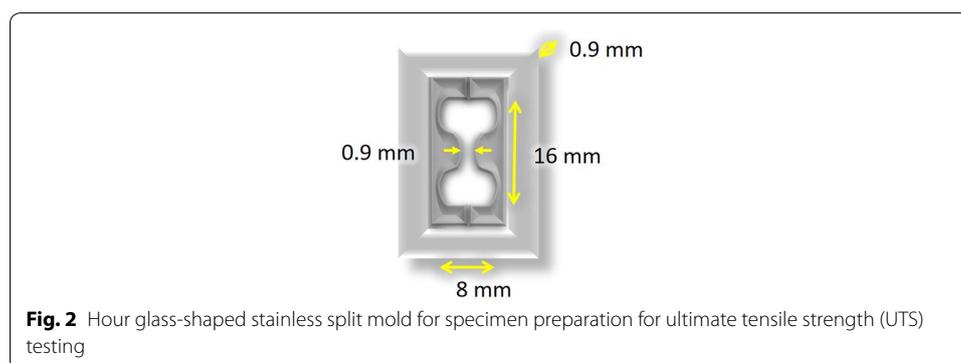
Data are the mean ± S.D., MPa;  $n = 10$

Same capital superscript letters indicate that ultimate tensile strength values were not significantly different among each group in the same water storage period ( $p > 0.05$ )

Same small superscript letters indicate that ultimate tensile strength values were not significantly different among each adhesive in the same water storage period ( $p > 0.05$ )

ANOVA results could not be detected the significant difference for "storage period" (horizontal rows)

\* Not applicable because adhesive could not be hardened even after sufficient light exposure



### Changes in specimen weight after water storage

The results for the percentage change in weight after water storage are shown in Table 3. According to the 3-way ANOVA analysis and Tukey's post hoc test, a significant effect was found for all three factors ( $p = 0.000$ ). A significant interaction was also found among the three factors ( $p = 0.000$ ).

There was a significant increase in weight for both the "Original" and "Desiccated" Clearfil SE Protect specimens after storage for 7 days than storage for 1 h ( $p < 0.05$ ). The "Original" OptiBond XTR showed slightly increase in weight after 1-h storage; however, this rather decreased after 7 days in storage. On the other hand, for the "Desiccated" OptiBond XTR samples, the percentage weight increase was higher after 7 days than 1 h. No weight measurements were taken for either of the 1-SEAs (OptiBond All-In-One and G-Premio Bond) in their "Original" forms. The weight of the "Desiccated" G-Premio Bond specimen decreased after storage in water, whereas the weight of the "Desiccated" OptiBond All-In-One specimen increased after storage, both in a time-dependent manner.

**Table 3 Percentage increase in weight after storage in water**

Group	Adhesive	1 h	7 days	S/NS
Original	Clearfil SE Protect	0.8 ± 0.4 <sup>B,a</sup>	6.7 ± 0.3 <sup>A,a</sup>	S
	OptiBond XTR	0.7 ± 0.7 <sup>B,a</sup>	-1.4 ± 1.5 <sup>B,b</sup>	S
	OptiBond All-In-One	NA*	NA*	-
	G-Premio Bond	NA*	NA*	-
Desiccated	Clearfil SE Protect	0.9 ± 0.3 <sup>B,a</sup>	5.9 ± 0.3 <sup>B,a</sup>	S
	OptiBond XTR	1.0 ± 0.3 <sup>AB,a</sup>	3.0 ± 1.0 <sup>C,a</sup>	S
	OptiBond All-In-One	2.2 ± 0.6 <sup>A</sup>	8.1 ± 0.7 <sup>A</sup>	S
	G-Premio Bond	-0.6 ± 0.2 <sup>C</sup>	-8.1 ± 2.1 <sup>D</sup>	S

Data are the mean ± S.D., %; *n* = 10

Same capital superscript letters indicate that ultimate tensile strength values were not significantly different among each group in the same water storage period (*p* > 0.05)

Same small superscript letters indicate that ultimate tensile strength values were not significantly different among each adhesive in the same water storage period (*p* > 0.05)

S, detection of the significant difference between 1 h and 7 days (*p* > 0.05); NS, not detection of the significant difference between 1 h and 7 days (*p* > 0.05); NA, not applicable because adhesive could not be hardened even after sufficient light exposure

## Discussion

The purpose of this study was to evaluate the influence of solvent evaporation on the mechanical properties of commercially available 2-SEAs and 1-SEAs. The mechanical property was measured in terms of ultimate micro-tensile strength (UTS), a variable that is frequently used to evaluate the performance of an adhesive in terms of its tensile bond strength with the local tissue [14, 15], which differs to the Young's modulus or the nano-indentation value [16].

Two representative 2-SEAs were compared in this study: Clearfil SE Protect and OptiBond XTR. The Clearfil SE Protect contains the same proprietary self-etching, light-curing technology as the "gold-standard" Clearfil SE Bond, which has been well described in the literature [17], but the Clearfil SE Protect has some additional ingredients: an MDPB antibacterial monomer in the self-etching primer, and sodium fluoride and HEMA in the adhesive agent [18, 19]. HEMA has been reported to affect water absorption even after the adhesive is polymerized. In contrast, OptiBond XTR contains ethanol but does not contain HEMA. Our results found differences in the behaviors of these two adhesives.

The "Original" Clearfil SE Protect adhesive showed high UTS after 1-h storage in water, which decreased after 7 days of storage in water. Both the "Original" and "Desiccated" specimens were also significantly heavier after 7 days in water, presumably because the Clearfil SE Protect contains HEMA monomers, which are hydrophilic and would have encouraged water sorption. For the "Original" specimens, this increased water sorption may be correlated with its reduced mechanical strength at 7 days; however, this is not the same for the "Desiccated" specimens, which showed no difference in UTS at 1 h or 7 days. Desiccation at 37 °C might accelerate the hydrolysis of hydrophilic monomers, such as HEMA and MDP [20]. However, we surmise that the 2-week desiccation period at 37 °C might have led to some deterioration of the adhesive's composition, and this may explain the difference in the UTS at the 1-h time point. Yet, without desiccation, this reduction in UTS in response to water sorption would, by extension,

suggest a similar reduction in bond strength of the adhesive to the tooth substrate as it was exposed to water over time [8]. Collectively, these results point to the need for the optimization of solvent evaporation for specific adhesive compounds.

Previous studies have shown that, although insufficient air-drying of Clearfil SE Bond primer can result in lower UTS of the Clearfil SE Bond adhesive, the UTS of the pure adhesive is very high [14, 21], presumably because of its high filler loading and polymerization efficacy [22]. It has also been reported that the adhesive layer thickness of the Clearfil SE Bond (which is of similar composition to Clearfil SE Protect) does not affect the bond strength [23]. Our results therefore confirm that strong air-blowing is unnecessary after applying the Clearfil SE Protect adhesive.

According to the manufacturer's report, filler loading of OptiBond XTR is higher than that of Clearfil SE Bond. However, the UTS of the "Original" OptiBond XTR was lower than that of the Clearfil SE Protect after 1 h of water storage. Also, in contrast to the Clearfil SE Protect, the "Original" OptiBond XTR had a lower specimen weight after water storage for 1 week. This phenomenon can be explained in two ways: Unlike the Clearfil SE Protect, the OptiBond XTR agent contains ethanol in the adhesive agent. Therefore, the percentage of the residual unpolymerized monomer within the light-cured OptiBond XTR specimens might be larger than that of Clearfil SE Protect. In addition, because OptiBond XTR does not contain HEMA, the water absorption of the OptiBond XTR is likely to be lower than that of Clearfil SE Protect. For the OptiBond XTR, the difference in the UTS between the "Original" and "Desiccated" specimens was smaller than that for Clearfil SE Protect. These results suggest that there is less likely to be a time-dependent degradation in bond strength, because the polymerized adhesives are less susceptible to water absorption [24]. The slight increase in the specimen weight may be due to the addition of glycerol-phosphate-dimethacrylate (GPDM) monomer [25].

For OptiBond XTR, desiccation did not improve the UTS. Others have shown that the addition of 10% to 20% ethanol to the experimental resin blends improves the degree of conversion [26]. Although the evaporation of the remaining ethanol in the OptiBond XTR might increase conversion, some ethanol remains trapped in the polymer and this would promote water sorption, which may lower the mechanical properties of the polymer [27, 28]. Therefore, in the clinic, excess evaporation of the OptiBond XTR adhesive might *negatively* affect the mechanical properties of polymer.

The "Original" OptiBond All-In-One adhesive did not hard even when light curing was used. This is probably due to the presence of water and other solvents, which inhibits the polymerization of the monomers in the adhesive resin [29]. The "Desiccated" OptiBond All-In-One adhesive, however, showed high UTS values. Almost all commercially available 1-SEAs contain water as a solvent to promote wetting of the substrate and improve the homogeneity of the liquid. Hydrophilic monomers enhance the wettability and infiltration of the hydrophobic resin monomers into the demineralized dentin matrix. However, insufficient air-blowing might result in the presence of residual solvents from the adhesive layer, and thereby lead to insufficient polymerization. According to the report of Fu and colleagues, the microtensile bond strength of OptiBond All-In-One applied to dentin was high (more than 75 MPa) if air-blowing was performed for more than 15 s. If air-blowing was omitted, all specimens failed before reaching the microtensile testing

phase [30]. Others show that solvent evaporation of OptiBond All-In-One using warm air-blowing ( $60\text{ }^{\circ}\text{C} \pm 2\text{ }^{\circ}\text{C}$ ) led to significantly higher bond strength than after using cold air-blowing ( $20\text{ }^{\circ}\text{C} \pm 1\text{ }^{\circ}\text{C}$ ) [31]. In the clinic, it is difficult to evaporate the solvent completely [32], because the relative humidity of the oral environment is higher than that of an experimental setting ( $23\text{ }^{\circ}\text{C}$ , 50%), and this is independent of whether the resin is applied to an incisal site or a molar site [33–35]. Resin bonding in a high humidity environment has reduced bond strength [36]. Therefore, it is recommended to control the intra-oral humidity using rubber dam isolation or an intra-oral vacuum device.

“Desiccated” G-Premio Bond, unlike the “Desiccated” OptiBond All-In-One, did not obtain adequate hardness even if with sufficient light curing, leaving a “gummy” specimen. In addition, whereas the weight of the OptiBond All-In-One specimens increased after 7 days of water storage, the weight of G-Premio Bond specimens decreased. Papadogiannis and colleagues [37] recently reported that Vickers hardness could not be measured in G-Premio Bond, since the material demonstrated a soft, gel consistency, contrary to all other 1-SEA (Adhese Universal, Ivoclar-Vivadent; All-Bond Universal, Bisco; Clearfil Universal Bond Quick, Kuraray Noritake Dental; Prelude One, Danville Materials; Scotchbond Universal, 3M ESPE). They also reported that the degree of C=C conversion of G-Premio Bond (67.2 DC%) was significantly lower than the all other 1-SEAs (77.7–82.0 DC%) [37]. G-Premio Bond has been reported the strong acidity (pH 1.5) than the other contemporary 1-SEAs, possibly associated with the presence of many acidic monomers (MDP, 4-MET, and MDTP) [30, 37]. Therefore, it can be suggested that the “gummy” G-Premio Bond specimen was resulted as the interference of polymerization. Ikeda et al. [14] reported that the UTS of G-Bond HEMA-free 1-SEA (GC) was significantly lower than that of HEMA-containing 1-SEA (Clearfil S<sup>3</sup> Bond, Kuraray Noritake Dental), despite the significantly higher amount of evaporation that occurs by air-drying for the G-Bond than the Clearfil S<sup>3</sup> Bond. They also observed numerous droplets in the G-Bond, regardless of the length of air-drying (0 s, 5 s, and 10 s). These droplets represent a phase-separation between water and the other adhesive ingredients [38, 39]. The authors explained that, because longer air-drying did not eliminate the droplets and thus the volatile parts of the adhesive, the droplets were likely encapsulated within the adhesive upon curing. The lower UTS and reduced specimen weight for the G-Premio Bond in our study supports their explanation; we surmise that similar droplets must have inhibited polymerization of adhesive and reduced its UTS. Tsujimoto et al. [40] recently reported numerous cracks in the fractured adhesive interfaces of G-Premio Bond adhesive after shear fatigue strength tests through SEM analysis. These defects at the adhesive interface might accelerate degradation at the resin-tooth adhesive interface. Therefore, further studies should be conducted to establish the relationship between residual solvents and bond durability.

## Conclusion

Based on the limitations imposed within this work, it can be concluded that the removal of residual solvent is important for polymerization of adhesives, particularly 1-SEAs. However, the presence of HEMA can negatively affect the mechanical properties of the cured adhesive, and desiccation for some types of adhesives can reduce the mechanical

strength after contact with water for an extended length of time. It is recommended that clinicians remove residual solvent to achieve optimal mechanical properties of the cured adhesive but perhaps consider the type of material and solvent before performing aggressive desiccation.

#### Abbreviations

1-SEA: one-step self-etch adhesive; 2-SEA: two-step self-etch adhesive; ANOVA: analysis of variance; Bis-GMA: 2,2-bis[4-(2-hydroxy-3-methacryloyloxypropoxy)phenyl]propane; GPDM: glycerol-phosphate-dimethacrylate; HEMA: 2-hydroxyethyl methacrylate; MDP: methacryloyloxydecyl dihydrogen phosphate; TEGDMA: triethylene glycol dimethacrylate; Tukey's HSD test: Tukey's honestly significant difference test; UDMA: urethane dimethacrylate; UTS: ultimate tensile strength.

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#### Authors' contributions

AK designed the study, prepared the specimens, and carried out the experimental measurements. AH carried out the experimental measurements and performed data analysis. HA, MK and YN carried out the experimental measurements. TM mainly reviewed the paper critically for content and approved it for submission. All authors read and approved the final manuscript.

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#### Availability of data and materials

The datasets generated during and/or analyzed during the current study are not publicly available, but are available from the corresponding author on reasonable request.

#### Ethics approval and consent to participate

This article does not contain any studies with human participants or animals performed by any of the authors.

#### Consent for publication

Not applicable.

#### Competing interests

The authors declare that they have no competing interests.

#### Author details

<sup>1</sup> Department of Operative Dentistry, Endodontology, and Periodontology, School of Dentistry, Matsumoto Dental University, 1780 Gobara Hirooka, Shiojiri, Nagano 399-0781, Japan. <sup>2</sup> Department of Operative Dentistry, Cariology and Pulp Biology, Tokyo Dental College, 2-9-18 Kanda-Misakicho, Chiyoda-ku, Tokyo 101-0061, Japan. <sup>3</sup> ABO Dental Clinic, 3-76-6 Keyaki-dai, Sanda, Hyogo 669-1321, Japan.

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